ENERGY CHANGES AND RATES OF REACTION

Lesson 10 Quiz: Reaction Mechanisms

Consider the reaction mechanism in the box for all questions.

1. What are the reactants for the overall chemical equation?

step 1 $A + B \rightarrow C$

step 2 $C + D \rightarrow E$

step 3 $E \rightarrow F + D$

step 4 $F + B \rightarrow G$

(fast)

(slow)

(fast)

(fast)

- A. E + 2D
- B. A + F
- C. A + 2B
- D. E + B
- 2. What are the products for the overall chemical equation?
 - A. G
 - B. C + D
 - C.C+D+G
 - D. 2E
- 3. What are the reaction intermediates?
 - A. C, E, F, D
 - B. D, E
 - C. C, D, G
 - D. C, E, F
- 4. What is the catalyst in the mechanism?
 - A. A
 - B. B
 - C. C
 - D. D
 - E. E
 - F. F
- 5. Which step is the rate-determining step?
 - A. step 1
 - B. step 2
 - C. step 3
 - D. step 4

ANSWERS

- 1. C
- 2. A
- 3. D
- 4. D
- 5. B